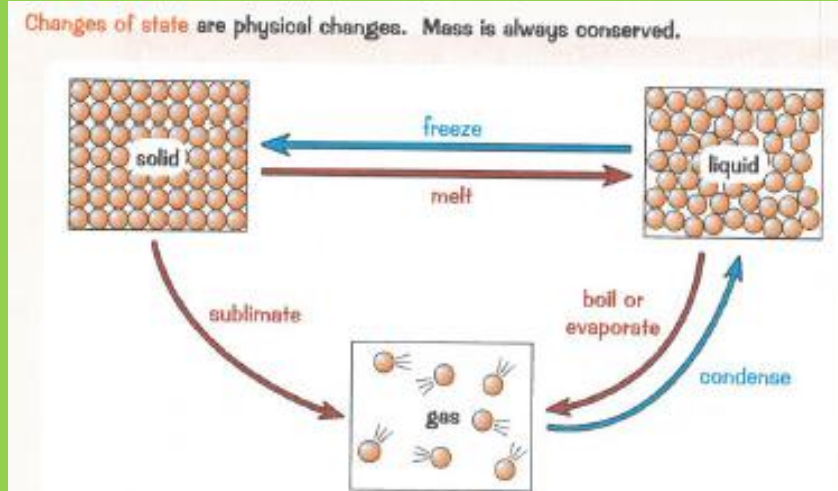


## Key Vocabulary

Internal energy	Total energy stored by the particles that make up a system.
Specific latent heat	The energy required to change 1kg of a substance with no change in temperature.
Specific heat capacity	The amount of energy needed to raise the temperature of 1kg of the substance by 1°C.
Specific latent heat of fusion	The specific latent heat of changing between a solid and a liquid.
Specific latent heat of vaporisation	The specific latent heat of changing between a liquid and a gas.

## Changes of state

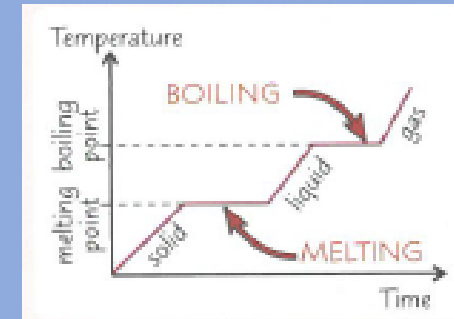


## Particle model

	Particle arrangement	Forces between particles	Distance between particles	Particle motion
<b>SOLID</b>	Regular, fixed	Strong	Very small	Vibration only
<b>LIQUID</b>	Irregular	Weak	Small	Slow
<b>GAS</b>	Irregular	Very weak	Large	Fast

Density decreases ↓

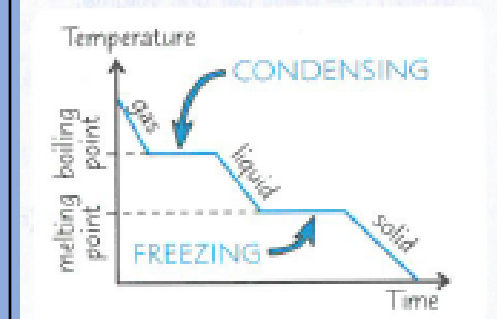
## Heating graphs



1- Bonds between particles are broken

2- Internal energy increases- energy is transferred to particles potential energy store.

## Cooling graphs



1- Stronger bonds form between particles.

2- Internal energy decreases- energy is transferred away from particles potential energy stores.

## Calculating density

**DENSITY** — mass per unit volume.

density (kg/m<sup>3</sup>)

$$\rho = \frac{m}{V}$$

mass (kg)

volume (m<sup>3</sup>)